

Note on Algebraic Structure Count

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Three recursion relations for the algebraic structure count are deduced, which are closely analogous to the well-known recursion relation for the number of Kekulé structures. An efficient graphical method for the calculation of algebraic structure count is proposed.

It is a long known result [1] that the determinant of the adjacency matrix (**A**) of a benzenoid graph is related to the number of the Kekulé structures (*K*) of the corresponding molecule as

$$\det \mathbf{A} = (-1)^{n/2} K^2. \quad (1)$$

For non-benzenoid (but alternant) systems, (1) is no more applicable and has to be replaced by

$$\det \mathbf{A} = (-1)^{n/2} L^2 \quad (2)$$

with *L* being a certain integer [2].

In the above equations and later on, *n* denotes the number of vertices of the molecular graph. It may be assumed that *n* is even since otherwise *K* = *L* = 0.

The quantity *L* has been called [3, 4] the “algebraic structure count” (or “corrected structure count” [5, 6]). Its relation to the Kekulé structures has been extensively examined [6–8]. The algebraic structure count plays an important role in the topological theory of non-benzenoid conjugated molecules [6, 9–12].

Let *e* be an (arbitrary) edge of the molecular graph *G*. Then *G* – *e* will denote the subgraph obtained by deleting this edge from *G*. In addition, *G* – (*e*) will denote the subgraph obtained from *G* by deletion of the two vertices which are incident to *G*. Hence, if *G* has *n* vertices, then *G* – *e* and *G* – (*e*) have *n* and *n* – 2 vertices, respectively.

The number of Kekulé structures of a molecular graph can be easily calculated by means of the recursion formula [6]

$$K(G) = K(G - e) + K(G - (e)). \quad (3)$$

On the other hand, no analogous regularity has been previously observed for the algebraic structure

count. The aim of the present paper is to show that *L*(*G*), *L*(*G* – *e*) and *L*(*G* – (*e*)) conform to one of the following three relations:

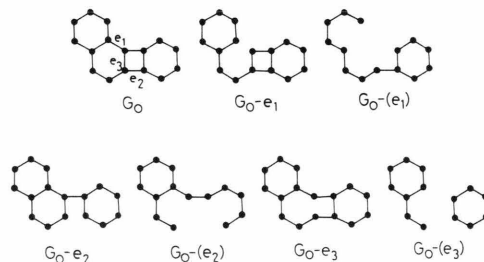
$$L(G) = L(G - e) + L(G - (e)), \quad (4)$$

$$L(G) = L(G - e) - L(G - (e)), \quad (5)$$

$$L(G) = L(G - (e)) - L(G - e) \quad (6)$$

which, of course, closely resemble (3).

As a matter of fact, (4), (5) and (6) may apply in the case of alternant non-benzenoid systems. This is illustrated by the example of the molecular graph *G*₀, whose three edges are labelled by *e*₁, *e*₂ and *e*₃. The corresponding subgraphs *G*₀ – *e*_{*i*} and *G*₀ – (*e*_{*i*}), *i* = 1, 2, 3, are given as follows:



Direct calculation shows that *L*(*G*₀) = 4 (see later), whereas *L*(*G*₀ – *e*₁) = *L*(*G*₀ – (*e*₁)) = *L*(*G*₀ – (*e*₂)) = 2, *L*(*G*₀ – *e*₂) = 6, *L*(*G*₀ – *e*₃) = 0 and *L*(*G*₀ – (*e*₃)) = 4. Thus (4) holds in the case of the deletion of the edge *e*₁, (5) applies for the edge *e*₂, and (6) must be used if the edge *e*₃ is being deleted.

In order to deduce (4)–(6), recall that the characteristic polynomial $\Phi(G)$ of the graph *G* is defined via

$$\Phi(G) = \Phi(G, x) = \det(x \mathbf{I} - \mathbf{A}), \quad (7)$$

where **I** is the unit matrix of dimension *n*. Then, because of (2),

$$\Phi(G, 0) = (-1)^{n/2} L^2. \quad (8)$$

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The characteristic polynomial conforms to the relations [13]

$$\Phi(G) = \Phi(G-e) - \Phi(G-(e)) - 2 \sum_Z \Phi(G-Z), \quad (9)$$

where the summation embraces all cycles Z which contain the edge e , and [14, 15]

$$\begin{aligned} \Phi(G-v_r) \Phi(G-v_s) - \Phi(G) \Phi(G-v_r-v_s) \\ = \left(\sum_W \Phi(G-W) \right)^2, \end{aligned} \quad (10)$$

where v_r and v_s are two vertices of G and the r.h.s. summation goes over all paths W which connect the vertices v_r and v_s . If v_r and v_s are adjacent vertices (joined by the edge e), then (10) can be rewritten in the form

$$\begin{aligned} \Phi(G-v_r) \Phi(G-v_s) - \Phi(G) \Phi(G-(e)) \\ = \left(\Phi(G-(e)) + \sum_Z \Phi(G-Z) \right)^2. \end{aligned} \quad (11)$$

Setting $x = 0$ into (9) and having (8) in mind, we get

$$L(G)^2 = L(G-e)^2 + L(G-(e))^2 + 2S, \quad (12)$$

where

$$S = \sum_Z (-1)^{(z/2)-1} L(G-Z)^2 \quad (13)$$

and z is the size of the cycles Z . (Note that if Z is of size $4m+2$, then $(-1)^{(z/2)-1} = +1$, and if Z is of size $4m$, then $(-1)^{(z/2)-1} = -1$. Cycles of odd size cannot occur in the molecular graphs of alternant systems.)

Set now $x=0$ into (11). Since n is even, the graphs $G-v_r$ and $G-v_s$ have an odd number of vertices and therefore $\Phi(G-v_r, 0) = \Phi(G-v_s, 0) = 0$. Consequently, we get from (11)

$$L(G)^2 L(G-(e))^2 = [L(G-(e))^2 + S]^2. \quad (14)$$

Eliminating the auxiliary quantity S from (12) and (14), we arrive at an identity connecting $L(G)$, $L(G-e)$ and $L(G-(e))$:

$$\begin{aligned} 4L(G)^2 L(G-(e))^2 \\ = [L(G)^2 + L(G-(e))^2 - L(G-e)^2]^2. \end{aligned} \quad (15)$$

The above relation can be further transformed into

$$\begin{aligned} [L(G) - L(G-(e)) - L(G-e)] \\ \cdot [L(G) - L(G-(e)) + L(G-e)] \\ \cdot [L(G) + L(G-(e)) - L(G-e)] \\ \cdot [L(G) + L(G-(e)) + L(G-e)] = 0. \end{aligned} \quad (16)$$

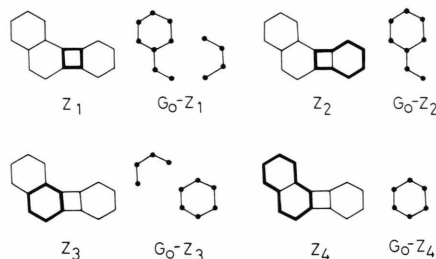
Since the last factor on the left-hand side of (16) is evidently non-zero, we conclude that the condition (15) implies the validity of either (4) or (5) or (6).

A detailed analysis of the relation (15) shows that formula (5) holds if the term $L(G-(e)) + S$ is negative (or zero). Formulas (4) and (6) hold when $L(G-(e)) + S$ is positive (or zero). Presently we are not able to determine the conditions which would make possible to predict whether (4) or (6) will apply for a given edge e of a given graph G .

In spite of their simple algebraic form, the recursion formulas (4)–(6) are of little value for the actual calculation of the algebraic structure count.

For practical purposes, (12) provides a much more efficient method of calculation of the algebraic structure count.

The use of (12) will be illustrated by the example of the graph G_0 . The edge which will be deleted is e_3 . There are four cycles in G_0 which contain e_3 . They will be labelled by Z_1 , Z_2 , Z_3 and Z_4 . The corresponding cycle-deleted subgraphs are given as follows:



We already know that $L(G_0-e_3) = 0$ and $L(G_0-(e_3)) = 4$. A simple calculation shows that in addition $L(G_0-Z_1) = L(G_0-Z_2) = L(G_0-Z_3) = L(G_0-Z_4) = 2$. Since $z_1 = 4$, $z_2 = 8$, $z_3 = 6$ and $z_4 = 10$, we have

$$L(G_0)^2 = 0^2 + 4^2 + 2(-2^2 - 2^2 + 2^2 + 2^2) = 16$$

and therefore $L(G_0) = 4$.

- [1] M. J. S. Dewar and H. C. Longuet-Higgins, *Proc. Roy. Soc. London* **A214**, 482 (1952).
- [2] For the theory of the determinant of the adjacency matrix of a molecular graph see: A. Graovac and I. Gutman, *Math. Chem. Mülheim* **6**, 49 (1979).
- [3] C. F. Wilcox, *Tetrahedron Letters* 795 (1968).
- [4] C. F. Wilcox, *J. Amer. Chem. Soc.* **91**, 2732 (1969).
- [5] W. C. Herndon, *Tetrahedron* **29**, 3 (1973).
- [6] For review of Kekulé structures, algebraic structure count and related topics see: W. C. Herndon, *J. Chem. Educ.* **51**, 10 (1974).
- [7] A. Graovac, I. Gutman, N. Trinajstić, and T. Živković, *Theoret. Chim. Acta* **26**, 67 (1972).
- [8] D. Cvetković, I. Gutman, and N. Trinajstić, *J. Chem. Phys.* **61**, 2700 (1974).
- [9] The topological theory of conjugated molecules is exposed in: A. Graovac, I. Gutman, and N. Trinajstić, *Topological Approach to the Chemistry of Conjugated Molecules*, Springer-Verlag, Berlin 1977.
- [10] C. F. Wilcox, *Croat. Chem. Acta* **47**, 87 (1975).
- [11] I. Gutman, N. Trinajstić, and C. F. Wilcox, *Tetrahedron* **31**, 143 (1975).
- [12] C. F. Wilcox, I. Gutman, and N. Trinajstić, *Tetrahedron* **31**, 147 (1975).
- [13] E. Heilbronner, *Helv. Chim. Acta* **36**, 170 (1953).
- [14] C. A. Coulson and H. C. Longuet-Higgins, *Proc. Roy. Soc. London* **A192**, 16 (1947).
- [15] For a recent application of Eq. (10) see: I. Gutman, *Z. Naturforsch.* **36a**, 1112 (1981).